

PHYSICS DATA SHEET

Constants

| | |
|---|---|
| Acceleration due to gravity near Earth | $ \vec{a}_g = 9.81 \text{ m/s}^2$ |
| Gravitational constant | $G = 6.67 \times 10^{-11} \text{ N} \cdot \text{m}^2/\text{kg}^2$ |
| Radius of the Earth | $r_E = 6.37 \times 10^6 \text{ m}$ |
| Mass of the Earth | $M_E = 5.97 \times 10^{24} \text{ kg}$ |
| Elementary Charge | $e = 1.60 \times 10^{-19} \text{ C}$ |
| Coulomb's Law Constant .. | $k = 8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2$ |
| Electron Volt | $1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$ |
| Index of Refraction of Air .. | $n = 1.00$ |
| Speed of Light in Vacuum .. | $c = 3.00 \times 10^8 \text{ m/s}$ |
| Planck's Constant | $h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$ $h = 4.14 \times 10^{-15} \text{ eV} \cdot \text{s}$ |
| Atomic Mass Unit | $u = 1.66 \times 10^{-27} \text{ kg}$ |

Physics Principles

- 0 Uniform motion ($\vec{F}_{net} = 0$)
- 1 Accelerated motion ($\vec{F}_{net} \neq 0$)
- 2 Uniform circular motion (\vec{F}_{net} is radially inward)
- 3 Work-energy theorem
- 4 Conservation of momentum
- 5 Conservation of energy
- 6 Conservation of mass-energy
- 7 Conservation of charge
- 8 Conservation of nucleons
- 9 Wave-particle duality

Prefixes Used with SI Units

| Prefix | Symbol | Exponential Value |
|-------------|-------------|-------------------|
| atto | a | 10^{-18} |
| femto | f | 10^{-15} |
| pico | p | 10^{-12} |
| nano | n | 10^{-9} |
| micro | μ | 10^{-6} |
| milli | m | 10^{-3} |
| centi | c | 10^{-2} |
| deci | d | 10^{-1} |
| deka | da | 10^1 |
| hecto | h | 10^2 |
| kilo | k | 10^3 |
| mega | M | 10^6 |
| giga | G | 10^9 |
| tera | T | 10^{12} |

Particles

| | Charge | Mass |
|----------------------|-------------|-----------------------------------|
| Alpha Particle | $+2e$ | $6.65 \times 10^{-27} \text{ kg}$ |
| Electron | $-1e$ | $9.11 \times 10^{-31} \text{ kg}$ |
| Proton | $+1e$ | $1.67 \times 10^{-27} \text{ kg}$ |
| Neutron | 0 | $1.67 \times 10^{-27} \text{ kg}$ |

First-Generation Fermions

| | Charge | Mass |
|--|-----------------------|----------------------------|
| Electron | $-1e$ | $0.511 \text{ MeV}/c^2$ |
| Positron | $+1e$ | $0.511 \text{ MeV}/c^2$ |
| Electron neutrino, ν | 0 | $< 50 \text{ eV}/c^2$ |
| Electron antineutrino, $\bar{\nu}$ | 0 | $< 50 \text{ eV}/c^2$ |
| Up quark, u | $+\frac{2}{3}e$ | $\sim 5 \text{ MeV}/c^2*$ |
| Anti-up quark, \bar{u} | $-\frac{2}{3}e$ | $\sim 5 \text{ MeV}/c^2*$ |
| Down quark, d | $-\frac{1}{3}e$ | $\sim 10 \text{ MeV}/c^2*$ |
| Anti-down quark, \bar{d} | $+\frac{1}{3}e$ | $\sim 10 \text{ MeV}/c^2*$ |

*Current models seem to suggest a significantly lower mass of these quarks than those in this table

EQUATIONS

Kinematics

$$\vec{v}_{\text{ave}} = \frac{\Delta \vec{d}}{\Delta t}$$

$$\vec{a}_{\text{ave}} = \frac{\Delta \vec{v}}{\Delta t}$$

$$\vec{d} = \vec{v}_i t + \frac{1}{2} \vec{a} t^2$$

$$|\vec{v}_c| = \frac{2\pi r}{T}$$

$$\vec{d} = \vec{v}_f t - \frac{1}{2} \vec{a} t^2$$

$$\vec{d} = \left(\frac{\vec{v}_f + \vec{v}_i}{2} \right) t$$

$$v_f^2 = v_i^2 + 2ad$$

$$|\vec{a}_c| = \frac{v^2}{r} = \frac{4\pi^2 r}{T^2}$$

Dynamics

$$\vec{a} = \frac{\vec{F}_{\text{net}}}{m}$$

$$|\vec{F}_f| = \mu |\vec{F}_N|$$

$$\vec{F}_s = -k\vec{x}$$

$$|\vec{F}_g| = \frac{Gm_1 m_2}{r^2}$$

$$|\vec{g}| = \frac{Gm}{r^2}$$

$$\vec{g} = \frac{\vec{F}_g}{m}$$

Momentum and Energy

$$\vec{p} = m\vec{v}$$

$$\vec{F} \Delta t = m \Delta \vec{v}$$

$$W = |\vec{F}| |\vec{d}| \cos \theta$$

$$W = \Delta E$$

$$P = \frac{W}{t}$$

$$E_k = \frac{1}{2} mv^2$$

$$E_p = mgh$$

$$E_p = \frac{1}{2} kx^2$$

Waves

$$T = 2\pi \sqrt{\frac{m}{k}}$$

$$T = 2\pi \sqrt{\frac{l}{g}}$$

$$T = \frac{1}{f}$$

$$v = f\lambda$$

$$f_o = \left(\frac{v}{v \mp v_s} \right) f_s$$

$$m = \frac{h_i}{h_o} = \frac{-d_i}{d_o}$$

$$\frac{1}{f} = \frac{1}{d_o} + \frac{1}{d_i}$$

$$\frac{\sin \theta_1}{\sin \theta_2} = \frac{\lambda_1}{\lambda_2} = \frac{v_1}{v_2} = \frac{n_2}{n_1}$$

$$\lambda = \frac{d \sin \theta}{n}$$

$$\lambda = \frac{dx}{nl}$$

Electricity and Magnetism

$$|\vec{F}_e| = \frac{kq_1 q_2}{r^2}$$

$$|\vec{E}| = \frac{kq}{r^2}$$

$$\vec{E} = \frac{\vec{F}_e}{q}$$

$$|\vec{E}| = \frac{\Delta V}{\Delta d}$$

$$\Delta V = \frac{\Delta E}{q}$$

$$I = \frac{q}{t}$$

$$|\vec{F}_m| = Il_{\perp} |\vec{B}|$$

$$|\vec{F}_m| = qv_{\perp} |\vec{B}|$$

$$V = Bvl \sin \theta$$

$$V = IR$$

Atomic Physics

$$W = hf_o$$

$$E_{k \text{ max}} = q_e V_{\text{stop}}$$

$$E = hf = \frac{hc}{\lambda}$$

$$N = N_o \left(\frac{1}{2} \right)^n$$

Quantum Mechanics and Nuclear Physics

$$\Delta E = \Delta mc^2$$

$$p = \frac{h}{\lambda}$$

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

$$E = pc$$

$$\Delta \lambda = \frac{h}{mc} (1 - \cos \theta)$$

Trigonometry and Geometry

$$\sin \theta = \frac{\text{opposite}}{\text{hypotenuse}}$$

$$\cos \theta = \frac{\text{adjacent}}{\text{hypotenuse}}$$

$$\tan \theta = \frac{\text{opposite}}{\text{adjacent}}$$

$$c^2 = a^2 + b^2$$

$$\frac{a}{\sin A} = \frac{b}{\sin B} = \frac{c}{\sin C}$$

$$c^2 = a^2 + b^2 - 2ab \cos C$$

Line

$$m = \frac{\Delta y}{\Delta x}$$

$$y = mx + b$$

Area

$$\text{Rectangle} = lw$$

$$\text{Triangle} = \frac{1}{2} ab$$

$$\text{Circle} = \pi r^2$$

Circumference

$$\text{Circle} = 2\pi r$$

Graphing Calculator Window Format

$$x : [x_{\text{min}}, x_{\text{max}}, x_{\text{scl}}]$$

$$y : [y_{\text{min}}, y_{\text{max}}, y_{\text{scl}}]$$

Periodic Table of the Elements

| | | | | | | | | | | | | | | | | | | |
|---|--|---|---|--|--|--|---|--|--|---|---|--|--|--|---|--|---|--|
| 1 H <small>1.01 hydrogen</small> | | | | | | | | | | | | | | | | | 2 He <small>4.00 helium</small> | |
| 3 Li <small>6.94 lithium</small> | 4 Be <small>9.01 beryllium</small> | | | | | | | | | | | | 5 B <small>10.81 boron</small> | 6 C <small>12.01 carbon</small> | 7 N <small>14.01 nitrogen</small> | 8 O <small>16.00 oxygen</small> | 9 F <small>19.00 fluorine</small> | 10 Ne <small>20.18 neon</small> |
| 11 Na <small>22.99 sodium</small> | 12 Mg <small>24.31 magnesium</small> | | | | | | | | | | | | 13 Al <small>26.98 aluminium</small> | 14 Si <small>28.09 silicon</small> | 15 P <small>30.97 phosphorous</small> | 16 S <small>32.07 sulfur</small> | 17 Cl <small>35.45 chlorine</small> | 18 Ar <small>39.95 argon</small> |
| 19 K <small>39.10 potassium</small> | 20 Ca <small>40.08 calcium</small> | 21 Sc <small>44.96 scandium</small> | 22 Ti <small>47.87 titanium</small> | 23 V <small>50.94 vanadium</small> | 24 Cr <small>52.00 chromium</small> | 25 Mn <small>54.94 manganese</small> | 26 Fe <small>55.85 iron</small> | 27 Co <small>58.93 cobalt</small> | 28 Ni <small>58.69 nickel</small> | 29 Cu <small>63.55 copper</small> | 30 Zn <small>65.39 zinc</small> | 31 Ga <small>69.72 gallium</small> | 32 Ge <small>72.64 germanium</small> | 33 As <small>74.92 arsenic</small> | 34 Se <small>78.96 selenium</small> | 35 Br <small>79.90 bromine</small> | 36 Kr <small>83.80 krypton</small> | |
| 37 Rb <small>85.47 rubidium</small> | 38 Sr <small>87.62 strontium</small> | 39 Y <small>88.91 yttrium</small> | 40 Zr <small>91.22 zirconium</small> | 41 Nb <small>92.91 niobium</small> | 42 Mo <small>95.94 molybdenum</small> | 43 Tc <small>(98) technetium</small> | 44 Ru <small>101.07 ruthenium</small> | 45 Rh <small>102.91 rhodium</small> | 46 Pd <small>106.42 palladium</small> | 47 Ag <small>107.87 silver</small> | 48 Cd <small>112.41 cadmium</small> | 49 In <small>114.82 indium</small> | 50 Sn <small>118.71 tin</small> | 51 Sb <small>121.75 antimony</small> | 52 Te <small>127.60 tellurium</small> | 53 I <small>126.90 iodine</small> | 54 Xe <small>131.29 xenon</small> | |
| 55 Cs <small>132.91 cesium</small> | 56 Ba <small>137.33 barium</small> | 57-71 | 72 Hf <small>178.49 hafnium</small> | 73 Ta <small>180.95 tantalum</small> | 74 W <small>183.84 tungsten</small> | 75 Re <small>186.21 rhenium</small> | 76 Os <small>190.23 osmium</small> | 77 Ir <small>192.22 iridium</small> | 78 Pt <small>195.08 platinum</small> | 79 Au <small>196.97 gold</small> | 80 Hg <small>200.59 mercury</small> | 81 Tl <small>204.38 thallium</small> | 82 Pb <small>207.21 lead</small> | 83 Bi <small>208.98 bismuth</small> | 84 Po <small>(209) polonium</small> | 85 At <small>(210) astatine</small> | 86 Rn <small>(222) radon</small> | |
| 87 Fr <small>(223) francium</small> | 88 Ra <small>(226) radium</small> | 89-103 | 104 Rf <small>(261) rutherfordium</small> | 105 Db <small>(262) dubnium</small> | 106 Sg <small>(266) seaborgium</small> | 107 Bh <small>(264) bohrium</small> | 108 Hs <small>(277) hassium</small> | 109 Mt <small>(268) meitnerium</small> | 110 Ds <small>(271) darmstadtium</small> | 111 Rg <small>(272) roentgenium</small> | 112 Uub <small>(285) ununbium</small> | 113 Uut <small>(284) ununtrium</small> | 114 Uuq <small>(289) ununquadium</small> | 115 Uup <small>(288) ununpentium</small> | 116 Uuh <small>(292) ununhexium</small> | 117 Uus <small>(?) ununseptium</small> | 118 Uuo <small>(294) ununoctium</small> | |

| | | | | | | | | | | | | | | |
|---|---|--|---|---|--|--|--|--|--|--|---|---|---|--|
| 57 La <small>138.91 lanthanum</small> | 58 Ce <small>140.12 cerium</small> | 59 Pr <small>140.91 praseodymium</small> | 60 Nd <small>144.24 neodymium</small> | 61 Pm <small>(145) promethium</small> | 62 Sm <small>150.36 samarium</small> | 63 Eu <small>151.96 europium</small> | 64 Gd <small>157.25 gadolinium</small> | 65 Tb <small>158.93 terbium</small> | 66 Dy <small>162.50 dysprosium</small> | 67 Ho <small>164.93 holmium</small> | 68 Er <small>167.26 erbium</small> | 69 Tm <small>168.93 thulium</small> | 70 Yb <small>173.04 ytterbium</small> | 71 Lu <small>174.97 lutetium</small> |
| 89 Ac <small>(227) actinium</small> | 90 Th <small>232.04 thorium</small> | 91 Pa <small>231.04 protactinium</small> | 92 U <small>238.03 uranium</small> | 93 Np <small>(237) neptunium</small> | 94 Pu <small>(244) plutonium</small> | 95 Am <small>(243) americium</small> | 96 Cm <small>(247) curium</small> | 97 Bk <small>(247) berkelium</small> | 98 Cf <small>(251) californium</small> | 99 Es <small>(252) einsteinium</small> | 100 Fm <small>(257) fermium</small> | 101 Md <small>(258) mendelevium</small> | 102 No <small>(259) nobelium</small> | 103 Lr <small>(262) lawrencium</small> |